2N57/ VI / Analytical finshoumentation

TNST 09/06/15

**QP Code: 5087** 

(3 Hours)

Total Marks: 80

N	R	(1)	Question	No 1	is	compulsory.	

- (2) Answer any three questions from the remaining five questions.
- (3) Assume suitable data wherever necessary.

	<ul> <li>100 - 100 -</li></ul>			
1.	Answer	the	tol	lowing

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- (a) Compare classical and Instrumental techniques of chemical analysis.
- (b) Why is column temperature so critical in GC?
- (c) List any 4 characteristic properties of Raman lines.
- (d) State Beer-Lambert's law and justify it as a limiting law.
- (e) Determine resonance frequency of proton in  $H_0 = 14092$  G:  $I = \pm 1/2$ ;  $\mu = 1.41 \times 10^{-30}$  IG+ and  $h = 6.626 \times 10^{-34}$  J sec.
- (a) With a neatdiagram explain the working of Single beam UV-VIS 10 Spectrometer.
  - (b) Explain in detail the concept of Fluorescence and Phosphorescence. 10 State the factors that influence Fluorescence. Also explain the working of Double beam Filter Fluorometer with neat diagram.
- 3. (a) Describe working and application of GM counter with neat diagram, 10
  - (b) Explain in detail working of Atomic Absorption Spectrometer with 10 neat diagram.
- (a) What is meant by Raman effect? Draw and explain the construction 10
  of Raman spectrometer with applications.
  - (b) Explain the working of Gas Chromatograph with a neat diagram. Also 10 state its applications.
- (a) Explain the basic working principle of Mass spectroscopy. Also with 10
  a neat diagram explain principle and working of Time of Flight Mass
  Spectrometer.
  - (b) Explain the concept of Nuclear Magnetic Resonance. With a neat 10 diagram explain in brief NMR spectrometer.

## 6. Write short notes on

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- (a) Monochromators
- (b) Chemical shift in NMR
- (c) X-ray tube
- (d) Oxygen analyzers

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